### References

MAIN, P., FISKE, S. J., HULL, S. E., LESSINGER, L., GERMAIN, G., DECLERCQ, J.-P. & WOOLFSON, M. M. (1980). MULTAN80. A System of Computer Programs for the Automatic Solution of Crystal Structures from X-ray Diffraction Data. Univs. of York, England, and Louvain, Belgium.

Acta Cryst. (1991). C47, 1766-1767

## Structure of a 1,4-Dien-3-one- $6\alpha$ -hydroxy Steroid

### By Hiroshi Nakai

Shionogi Research Laboratories, Shionogi & Co. Ltd, Fukushima-ku, Osaka 553, Japan

(Received 10 January 1991; accepted 6 February 1991)

C(1)

C(2) C(3)

C(4)

C(5) C(6)

C(7) C(8)

C(9)

C(10)

C(11) C(12) C(13) C(14)

 $\dot{c}(15)$ 

C(16)

C(17) C(18)

C(19) C(20)

C(21)

C(22) Cl(23)

Cl(24)

O(25) O(26)

O(27) O(28)

O(29) C(30) O(31)

C(32)

C(33) C(34)

C(35)

O(36)

Abstract.  $9\alpha$ ,21-Dichloro- $6\alpha$ ,11 $\beta$ ,17 $\alpha$ -trihydroxy-16α-methyl-3,20-dioxopregna-1,4-dien-17-yl 2-furoate,  $C_{27}H_{30}Cl_2O_7$ ,  $M_r = 537.43$ , monoclinic,  $P2_1$ , a =11.524 (2), b = 15.751 (4), c = 7.975 (2) Å,  $\beta =$ V = 1347.7 (6) Å<sup>3</sup>, Z = 2,  $111.42(1)^{\circ}$ ,  $D_{r} =$  $1.324 \text{ Mg m}^{-3}$ ,  $\lambda(\mathrm{Cu}\; K\alpha) = 1.54178\; \mathrm{\AA},$  $\mu =$  $2.53 \text{ mm}^{-1}$ , F(000) = 564, T = 295 K, R = 0.047 forobserved reflections  $[F_o > 3\sigma(F_o)]$ . 2298 The molecules are linked by intermolecular hydrogen O(26)-HO(27)(1 - x,  $y - \frac{1}{2}$ , 1 - z) = bonds: 2.697 (5) [2.00 (6) for O...H] and O(31)-HO(26)  $(2 - x, \frac{1}{2} + y, 2 - z) = 2.895$  (6) Å [2.04 (8) Å].

Experimental. Colorless plate crystals obtained from methanol. Crystal of dimensions  $0.3 \times 0.3 \times 0.1$  mm. Rigaku AFC-5R diffractometer, graphite-monochromatized Cu  $K\alpha$ . Cell dimensions determined from  $2\theta$  angles for 25 reflections in the range  $26 < 2\theta < 46^{\circ}$ . Intensities measured up to  $\theta = 70^{\circ}$  in h = 13/14, k 0/19 and l = 9/0,  $\omega = 2\theta$  scans,  $\omega$ -scan width  $(2 + 0.2 \tan \theta)^{\circ}$ , three standard reflections monitored every 100 measurements showed no significant change. 2571 unique reflections measured, 2298 intensities observed  $[F_a \leq 3\sigma(F_a)]$  and one very strong reflection rejected], no absorption correction. Structures solved by direct methods with MULTAN84 (Main, Germain & Woolfson, 1984). H atoms located on a difference density map. Positional parameters of all atoms and anisotropic thermal parameters of non-H atoms refined by block-diagonal least squares. Temperature factor of each H atom equal to  $B_{eq}$  of the bonded atom.  $\Sigma(w|\Delta F|^2)$  minimized,  $w = 1/[\sigma^2(F_o) + 0.00146|F_o|^2]$ , w = 0 for 53 reflections with  $w^{1/2} |\Delta F| \ge 3$ . Final R =0.047, wR = 0.055, S = 1.1150. Max.  $\Delta/\sigma$  in the final cycle 0.03. The highest and lowest peaks in the final difference map are 0.6 and -0.5 e Å<sup>-3</sup>. Atomic scattering factors calculated by  $\sum [a_i \exp(-b_i \lambda^{-2} \sin^2 \theta)] +$ c (i = 1,...,4) (International Tables for X-ray CrystalTable 1. Atomic coordinates and equivalent isotropic temperature factors  $(A^2)$ 

OHSAWA, A., ARAI, H., OHNISHI, H., ITOH, T., KAIHOH, T., OKADA, M. & IGETA, H. (1985). J. Org. Chem. 50, 5520–5523. Rigaku Corporation (1985). RCRYSTAN. X-ray Analysis

YAO, J.-X., ZHENG, C.-D., QIAN, J.-Z., HAN, F.-S., GU, Y.-X. &

FAN, H.-F. (1985). SAP185. Institute of Physics, Chinese Academy of Sciences, Beijing, People's Republic of China.

Program System. Rigaku Corporation, Tokyo, Japan.

$$B_{\rm eq} = \frac{4}{3} \sum_i \sum_j \beta_{ij} \mathbf{a}_i \cdot \mathbf{a}_j.$$

| x          | у          | Z           | $B_{eq}$ |
|------------|------------|-------------|----------|
| 0.4496 (4) | 0.2944 (3) | 0.3006 (5)  | 4.1(1)   |
| 0.4305 (4) | 0.2601 (3) | 0.1399 (5)  | 4·5 (1)  |
| 0.4636 (5) | 0.1717 (3) | 0.1247 (5)  | 4.8 (1)  |
| 0.5123 (4) | 0.1232(3)  | 0.2894 (5)  | 4.2 (1)  |
| 0.5293 (3) | 0.1560 (3) | 0.4511 (5)  | 3.4 (1)  |
| 0.5800 (4) | 0.1035 (3) | 0.6214 (5)  | 3.7 (1)  |
| 0.6837 (4) | 0.1481 (3) | 0.7692 (5)  | 3.7 (1)  |
| 0.6607 (3) | 0.2419 (3) | 0.7940 (5)  | 3.2 (1)  |
| 0.6162 (3) | 0.2900 (3) | 0.6137 (4)  | 3.0 (1)  |
| 0.4958 (4) | 0.2467 (3) | 0.4755 (5)  | 3.5 (1)  |
| 0.6087 (3) | 0.3875 (2) | 0.6367 (5)  | 3.3 (1)  |
| 0.7241 (4) | 0.4244 (2) | 0.7837 (5)  | 3.4 (1)  |
| 0.7605 (4) | 0.3775 (3) | 0.9640 (5)  | 3.4 (1)  |
| 0.7766 (3) | 0.2827 (3) | 0.9289 (4)  | 3.3 (1)  |
| 0.8322 (4) | 0.2440 (3) | 1.1144 (5)  | 4.1 (1)  |
| 0.9221 (4) | 0.3136 (3) | 1.2277 (5)  | 4.2 (1)  |
| 0.8932 (4) | 0.3952 (3) | 1.1073 (5)  | 3.6 (1)  |
| 0.3848 (4) | 0.2431 (3) | 0.5421 (6)  | 4.3 (1)  |
| 0 6645 (4) | 0.3924 (3) | 1.0517 (5)  | 4.1 (1)  |
| 0.8997 (4) | 0.4774 (3) | 1.2155 (6)  | 4.6 (1)  |
| 0.8873 (6) | 0.5591 (4) | 1-1128 (8)  | 6.2 (2)  |
| 1.0559 (5) | 0.2841 (5) | 1.3026 (8)  | 6.9 (2)  |
| 0.7371 (1) | 0.2798     | 0-5149 (1)  | 3.79 (2) |
| 0.9575 (3) | 0.6477 (1) | 1.2487 (3)  | 10.9 (1) |
| 0.4511 (4) | 0.1391 (3) | -0.0219 (4) | 7.2 (2)  |
| 0.6146 (3) | 0.0219 (2) | 0.5876 (4)  | 5.4 (1)  |
| 0.4992 (2) | 0.4065 (2) | 0.6700 (4)  | 3.9 (1)  |
| 0.9020 (3) | 0.4765 (3) | 1.3662 (4)  | 6.2 (1)  |
| 0.9740 (2) | 0.4019 (2) | 1.0035 (4)  | 3.9 (1)  |
| 1.0879 (3) | 0.4338 (3) | 1.0767 (5)  | 3.9 (1)  |
| 1.1334 (3) | 0.4612 (3) | 1.2299 (4)  | 5.3 (1)  |
| 1.1485 (4) | 0.4369 (3) | 0.9464 (6)  | 4·3 (1)  |
| 1.1102 (5) | 0.4264 (6) | 0.7686 (7)  | 7.5 (2)  |
| 1.2104 (6) | 0.4453 (6) | 0.7177 (8)  | 8-3 (3)  |
| 1.3070 (5) | 0.4622 (4) | 0.8680 (7)  | 5.7 (2)  |
| 1.2707 (3) | 0.4586 (2) | 1.0116 (4)  | 4.5 (1)  |

*lography*, 1974, Vol. IV). Calculations performed on a VAX station 3100 computer. The final atomic coordinates and equivalent isotropic temperature factors are given in Table 1.\* Bond lengths and

© 1991 International Union of Crystallography

<sup>\*</sup> Lists of H-atom coordinates, anisotropic temperature factors of the non-H atoms and structure factors have been deposited with the British Library Document Supply Centre as Supplementary Publication No. SUP 53985 (19 pp.). Copies may be obtained through The Technical Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

| C(1)—C(2) 1                             | .333 (7)    | C(13)-C(14)  | 1.543   | (7)              |
|---|-------------|--|---------|------------------|
| C(1)-C(10)                              | .500 (7)    | C(13)-C(17)  | 1.564   | (7)              |
| C(2) - C(3)                             | ·460 (7)    | C(13) - C(19)  | 1.528   |                  |
| C(3) - C(4)                             | .444 (7)    | $C(14) \rightarrow C(15)$  | 1.509   | $\tilde{\alpha}$ |
| C(3) = O(25)                            |             | C(15) - C(16)  | 1.552   | (7)              |
| C(4) - C(5)                             | 1.236 (7)   | C(16) - C(17)  | 1.566   | (7)              |
| C(4) = C(3)                             |             | C(10) - C(17)  | 1.500   | (7)              |
| C(5) - C(6)                             | 1.515(7)    | C(10) - C(22)  | 1.509   | (9)              |
| C(5) - C(10)                            | 1.511 (7)   | C(17) - C(20)  | 1.543   | ()               |
| C(6) - C(7)                             | 1.511 (7)   | C(17)—O(29)  | 1.458   | (6)              |
| C(6)—O(26)                              | l·401 (6)   | C(20)—C(21)  | 1.504   | (8)              |
| C(7)—C(8)                               | 1.527 (7)   | C(20)—O(28)  | 1-192   | (7)              |
| C(8)—C(9)                               | l·538 (7)   | C(21)—Cl(24)   | 1.771   | (8)              |
| C(8)—C(14)                              | l·518 (7)   | O(29)—C(30)  | 1.325   | (6)              |
| C(9)-C(10)                              | l·577 (7)   | C(30)-O(31)  | 1.219   | (7)              |
| C(9) - C(11)                            | 1.553 (6)   | C(30) - C(32)  | 1.449   | (7)              |
| C(9) - Cl(23)                           | -841 (5)    | C(32)-C(33)  | 1.333   | à'n              |
| $C(10) \rightarrow C(18)$               | 1.555 (7)   | C(32) = O(36)  | 1.355   | (6)              |
| C(10) = C(10)                           | 1.530 (6)   | C(33) - C(34)  | 1.380   | (13)             |
| C(11) - C(12)                           | 1 3 3 0 (0) | C(34) = C(34)  | 1.221   | (13)             |
| C(11) = O(27)                           | (413 (3)    | C(34) - C(33)  | 1.351   | (11)             |
| C(12) - C(13)                           | 1.535 (7)   | C(35)-O(36)  | 1.330   | ()               |
|   | 124 7 (4)   | 0(12) 0(12)  | 0(17)   | 117 2 (4)        |
| C(2) - C(1) - C(10)                     | 124.7 (4)   | C(12) - C(13) - C(13)  | -C(17)  | 117.3 (4)        |
| C(1) - C(2) - C(3)                      | 120-7 (5)   | C(12) - C(13) - C(13)  | -C(19)  | 111-1 (4)        |
| C(2) - C(3) - C(4)                      | 117-1 (5)   | C(14)—C(13)—   | -C(17)  | 98-9 (4)         |
| C(2)—C(3)—O(25)                         | 122-1 (5)   | C(14)-C(13)-   | -C(19)  | 112.8 (4)        |
| C(4)C(3)O(25)                           | 120.9 (5)   | C(17)-C(13)-   | -C(19)  | 108.4 (4)        |
| C(3) - C(4) - C(5)                      | 123.0 (5)   | C(8)-C(14)   | C(13)   | 113.9 (4)        |
| C(4)-C(5)-C(6)                          | 121-8 (4)   | C(8)-C(14)-  | C(15)   | 118.2 (4)        |
| $\dot{c}(4) - \dot{c}(5) - \dot{c}(10)$ | 122.4 (4)   | $C(13) \rightarrow C(14)$  | -Èus    | 104.4 (4)        |
| C(6) = C(5) = C(10)                     | 115.7 (4)   | C(14) - C(15) - C(15 | -C(16)  | 103.9 (4)        |
| C(5) = C(5) = C(7)                      | 112.5 (4)   | C(15) - C(16) - C(16)  | -C(17)  | 106.0 (4)        |
| C(5) = C(0) = C(7)                      | 112 0 (4)   | C(15) = C(16)  | C(1)    | 112.4 (5)        |
| C(3) - C(0) - O(20)                     | 111.9 (4)   |  | -C(22)  | 112.4 (3)        |
| C(7) - C(6) - O(26)                     | 112.2 (4)   |  | -C(22)  | 117.0 (5)        |
| C(6) - C(7) - C(8)                      | 115.2 (4)   | C(13) - C(17) - C(17 | -C(16)  | 103.5 (4)        |
| C(7) - C(8) - C(9)                      | 111.7 (4)   | C(13) - C(17) - C(17)  | -C(20)  | 113.2 (4)        |
| C(7)—C(8)—C(14)                         | 110.6 (4)   | C(13)—C(17)—   | -O(29)  | 104.8 (4)        |
| C(9) - C(8) - C(14)                     | 111-1 (4)   | C(16)—C(17)—   | -C(20)  | 113.0 (4)        |
| C(8)—C(9)—C(10)                         | 110-4 (4)   | C(16)—C(17)—   | -O(29)  | 111.8 (4)        |
| C(8)-C(9)-C(11)                         | 113.0 (4)   | C(20)—C(17)—   | -O(29)  | 110.3 (4)        |
| C(8)-C(9)-Cl(23)                        | 108-1 (3)   | C(17)-C(20)-   | -C(21)  | 116.0 (4)        |
| C(10) - C(9) - C(11)                    | 115.9 (4)   | C(17)-C(20)-   | -O(28)  | 122.2 (5)        |
| C(10) - C(9) - Cl(23)                   | 105.9 (3)   | C(21)-C(20)-   | -O(28)  | 121.5 (5)        |
| C(11) - C(9) - C(23)                    | 102.7 (3)   | C(20) - C(21) -  | -CI(24) | 114.2 (5)        |
| C(1) - C(10) - C(5)                     | 112.0 (4)   | C(17) - O(29) -  | -C(30)  | 120.8 (4)        |
| C(1) = C(10) = C(9)                     | 110.9 (4)   | 0(29) - C(30) - C(30 | -0(31)  | 124.8 (5)        |
| C(1) = C(10) = C(18)                    | 106.2 (4)   | O(29) - C(30) - C(30 | -C(32)  | 110.9 (4)        |
| C(1) = C(10) = C(10)                    | 107.2 (4)   | O(21) - C(30)  | -C(32)  | 124.3 (5)        |
| C(3) = C(10) = C(19)                    | 107.5 (4)   | C(30) = C(30) =  | -C(32)  | 134.4 (4)        |
| C(0) = C(10) = C(10)                    | 112 4 (4)   | C(30) - C(32) - C(32)  | 0(24)   | 114.0 (4)        |
| C(3) = C(10) = C(18)                    | 113.0 (4)   | C(30) - C(32) - C(32)  | 0(30)   | 100 € (4)        |
| C(9) - C(11) - C(12)                    | 113-3 (3)   | C(33) - C(32) - C(32)  | -0(36)  | 107.3 (3)        |
| C(9) - C(11) - O(27)                    | 108-9 (3)   | C(32) - C(33) - C(33)  | -C(34)  | 107-3 (8)        |
| C(12) - C(11) - O(27)                   | 111-2 (3)   | C(33)—C(34)—   | -C(35)  | 106.9 (8)        |
| C(11) - C(12) - C(13)                   | 113.5 (4)   | C(34)—C(35)—   | -O(36)  | 109.7 (6)        |
| C(12) - C(13) - C(14)                   | 107.8 (4)   | C(32)-O(36)-   | -C(35)  | 106.5 (4)        |

Table 2. Bond lengths (Å) and angles (°)

angles are listed in Table 2. A perspective view of the molecule with atom labeling is presented in Fig. 1.



Fig. 1. Perspective view drawn by PLUTO (Motherwell & Clegg, 1978).

Related literature. The structure of the title compound reported here has been referred to in the chemical synthesis of 6-hydroxy corticosteroids (Terasawa & Okada, 1991).

The author thanks Drs Terasawa and Okada for supplying the crystals.

#### References

- MAIN, P., GERMAIN, G. & WOOLFSON, M. M. (1984). MULTAN84. A Computer Program for the Automatic Solution of Crystal Structures from X-ray Diffraction Data. Univs. of York, England, and Louvain, Belgium.
- MOTHERWELL, W. D. S. & CLEGG, W. (1978). PLUTO. Program for plotting molecular and crystal structures. Univ. of Cambridge, England.
- TERASAWA, T. & OKADA, T. (1991). Synth. Commun. 21, 307-317.

Acta Cryst. (1991). C47, 1767-1769

# Structure of a 1,4-Dien-3-one- $6\beta$ -hydroxy Steroid Methanol Solvate

By Hiroshi Nakai

Shionogi Research Laboratories, Shionogi & Co. Ltd, Fukushima-ku, Osaka 553, Japan

(Received 10 January 1991; accepted 6 February 1991)

 $9\alpha$ ,21-Dichloro- $6\beta$ ,11 $\beta$ ,17 $\alpha$ -trihydroxy-Abstract.  $16\alpha$ -methylpregna-1,4-diene-3,20-dione methanol solvate,  $C_{22}H_{28}Cl_2O_{5,\frac{1}{2}}CH_3OH$ ,  $M_r = 459.385$ , mono-

C2, a = 25.506(5), b = 8.371(1),clinic, c =15.035 (3) Å,  $\beta = 136.99$  (1)°, V = 2189.8 (8) Å<sup>3</sup>, Z =4,  $D_x = 1.393 \text{ Mg m}^{-3}$ ,  $\lambda(\text{Cu } K\alpha) = 1.54178 \text{ Å}$ ,  $\mu =$ 

0108-2701/91/081767-03\$03.00

© 1991 International Union of Crystallography